**Describing & Explaining trends in Electronegativity (Level 3) examiners tips: Read these please!**

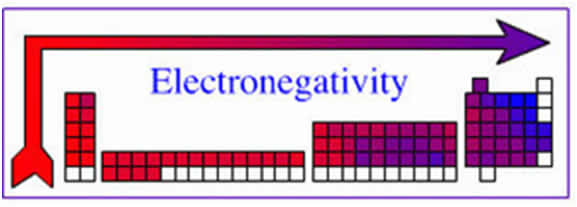
• always relate the electronegativity of atoms in a molecule to

**1)** the polar/non-polar bonds within a molecule

and

**2)** to the overall shape of the molecule

• see below a graph that summarises the trend in electronegativity



• generally electronegativity increases across a period and up a group

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